ANALYSIS OF SAFETY BEHAVIORS AND FAILURE MECHANISMS FOR LITHIUM-ION BATTERIES IN ABUSE TESTS



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Outline



- Key safety concerns in lithium-ion batteries (LIBs)
- Study of failure mechanism(s) in LIB abuse testing

Internal Short Circuit Simulation test

Overcharging test

Thermal Abuse Test

Summary





Key Safety Concern(s) in Lithium-ion Batteries

- Three major safety concerns while using a LIB.
 - Internal Short-Circuit → No effective way to completely avoid
 - Thermal Abuse → Some materials inside LIB are flammable
 - Overcharging → LIB materials are extremely active under high cell potential
- Emerging Safety Concern(s): Higher power requirement, higher power density, more severe environment, longer service life, etc.



Single Cell Pack Cell Phone 1-3 years life



6-12 Cells in a Pack Laptop 1-3 years life



100-5000 Cells in a System EV/HEV and Motive 5-15 years life

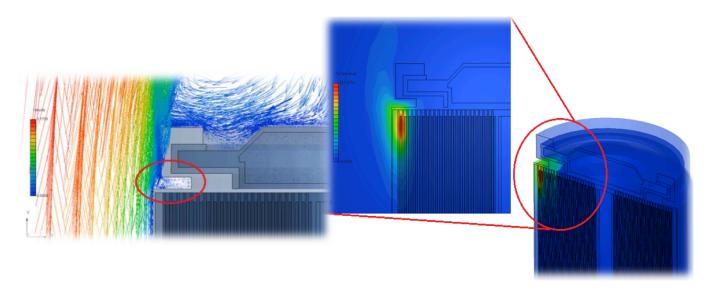


>1000 Cells in a System Energy Storage and Stationary 10-20 years life



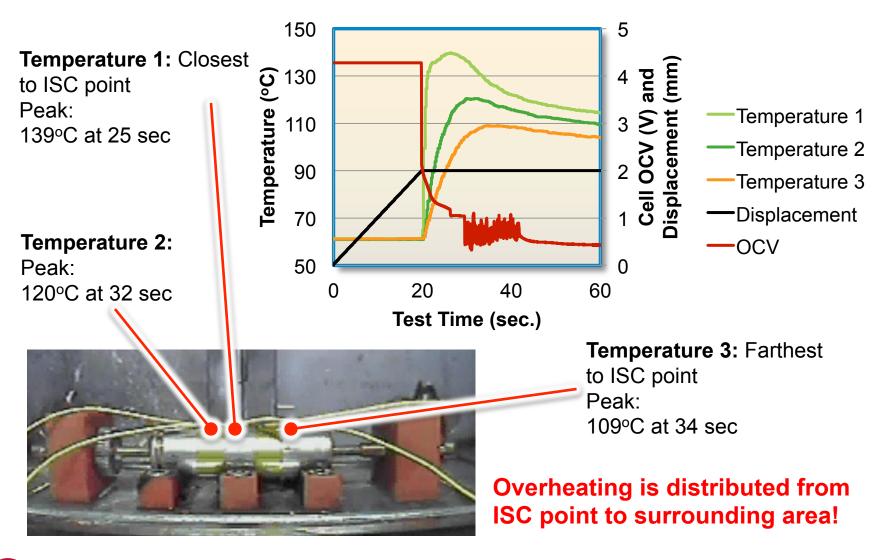
Key Characteristics of ISC Event

- Most ISC events start from localized heating, due to multiple factors, such as manufacturing flaws, mechanical abuse, etc.
- Once the ISC triggers, all safety designs outside of cells are not effective
- Most safety devices within single cell will be also bypassed (ex. fuse, PTC, CID, etc...)



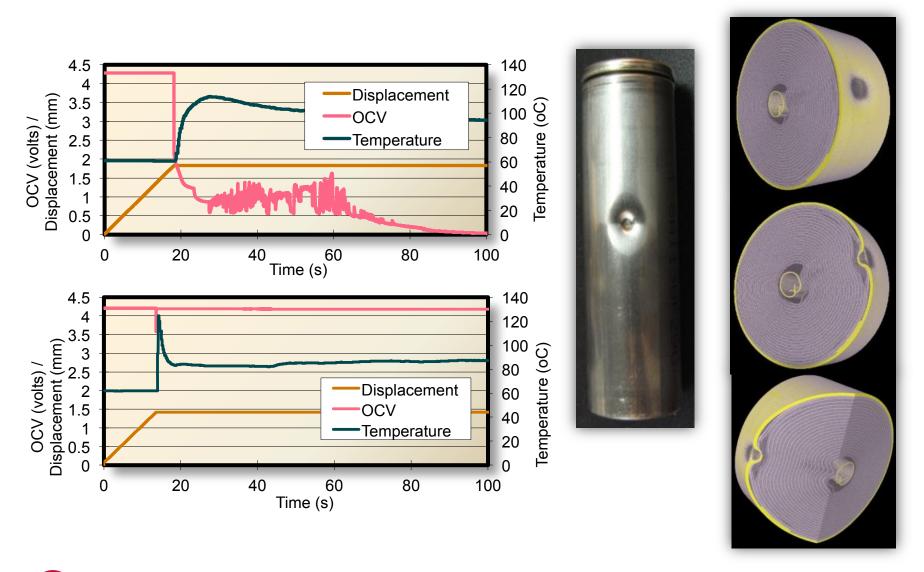


Behaviors Observed from ISC Test



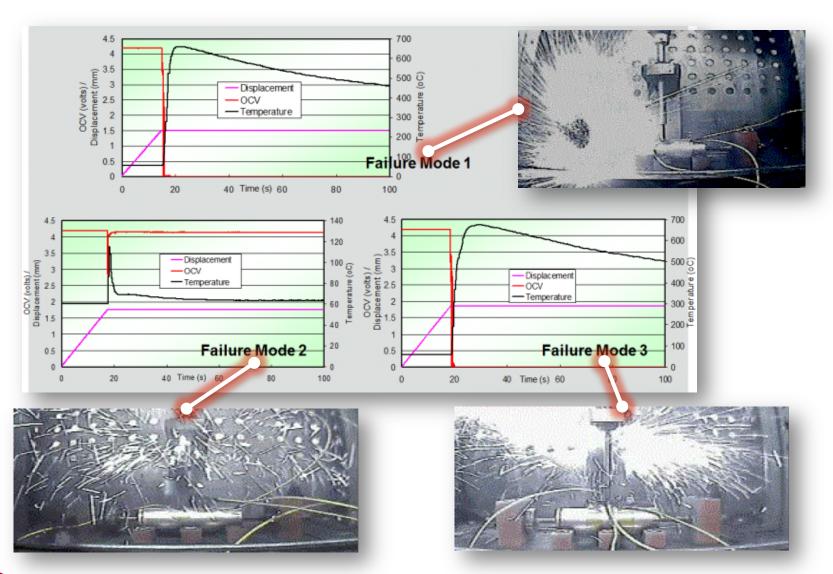


Observed Failure Modes via ISC test





Observed Failure Modes via ISC test (Cont.)



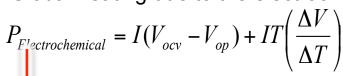


Typical Failure Mechanism in an ISC

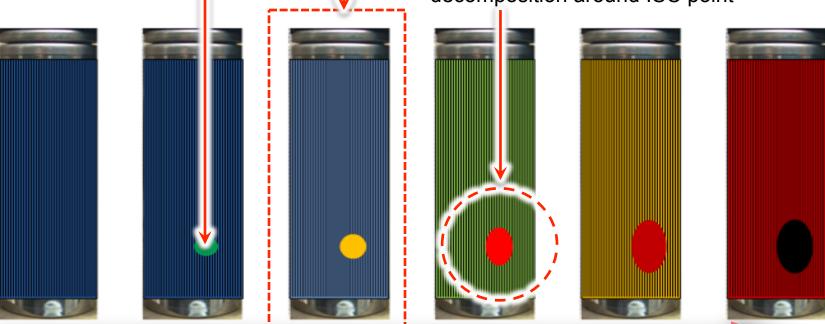
Global heating due to the electrochemical reaction

Localized over-heating by Joule Effect at ISC point

$$P_{Elect\,rical} = IV = I^2R$$



Local over-heating to trigger material decomposition around ISC point



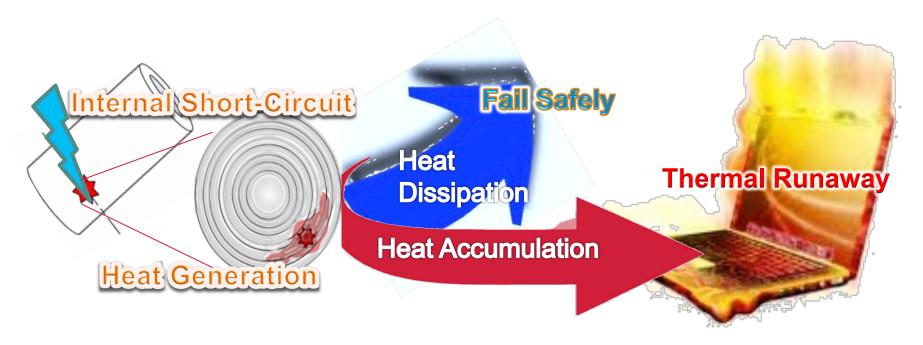
Time



Factors to Cause Thermal Runaway in an ISC Event

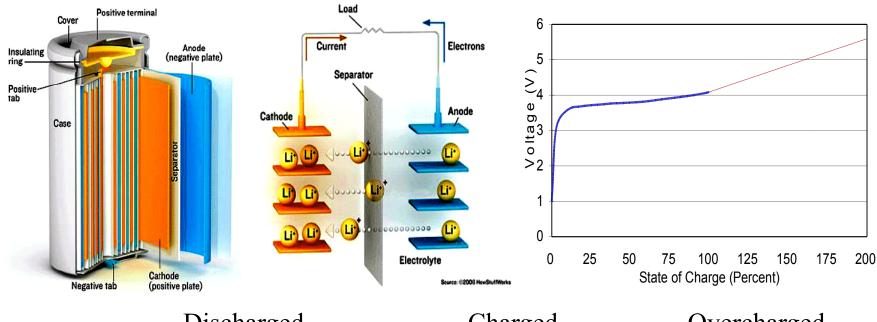
2 Key Factors to "Fail Safely" or "Thermal Runaway" in an ISC event:

- Heat Generation Rate
- Capability of the Cell Design to dissipate excess heat in short time



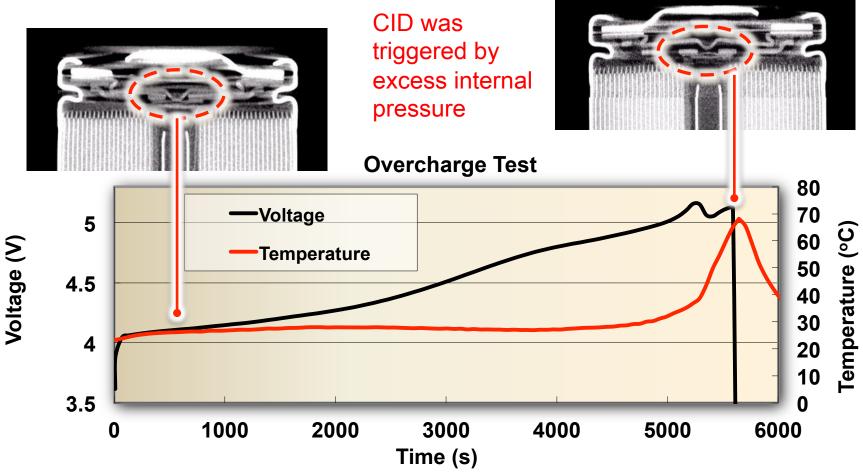


Safety Concern of LIB under Overcharge Condition





LiCoO₂ LIB Behavior(s) under Overcharge Test

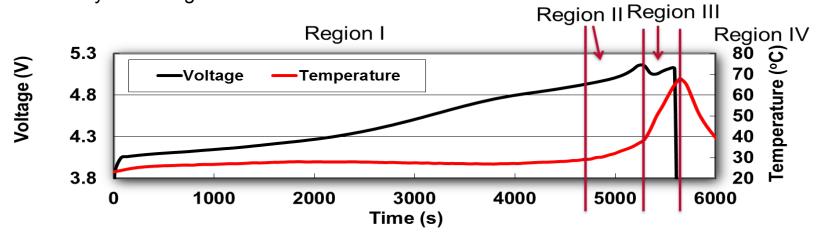


If CID is disabled, LiCoO₂ type LIB will usually result in thermal runaway after cell voltage is higher than 5.0 to 5.4V due to the abundant heat generated via the violent reaction of electrolyte material with electrodes



Typical Failure Mechanism in Overcharge Test

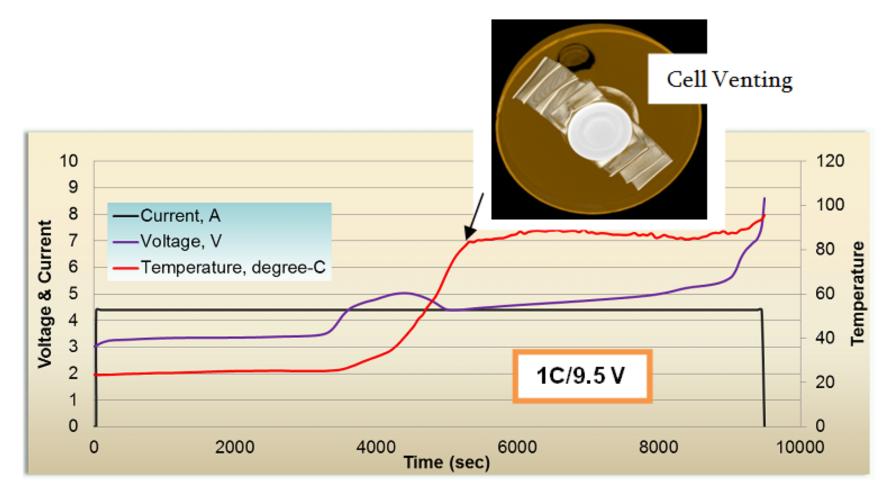
- Region I: Small temperature rise. Depends on cell anode-to-cathode balance design, lithium metal can be deposited on anode.
- ♠ Region II: Temperature begins to rise at a slow rate. Electrolyte oxidation can occur at cathode under high voltage. An increase in DC resistance found at around 80 to 95% charge range (i.e. Li_xCoO₂, x=0.2-0.05), but return to a lower value at >95% charge (i.e. x<0.05).</p>
- ♦ Region III: Temperature begins to rise more quickly. Cell voltage drops and recovers due to the change of cell DC resistance.
- ♠ Region IV: Cell can fail safely with the aid of CID or other safety devices. However, if there is no effective protection device activated, LiCoO₂ type LIB is like to result in thermal runaway under region IV



Source: R. A. Leising et al., J. of The Electrochemical Society, 148 (8) A838-A844 (2001)



LFP LIB Behavior(s) under Overcharge Test



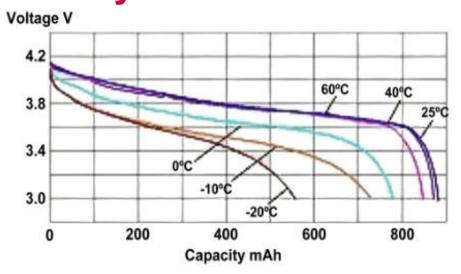
LFP type LIB will not exhibit thermal runaway even under high cell voltage (>9V), because LFP will not oxidize electrolyte materials.

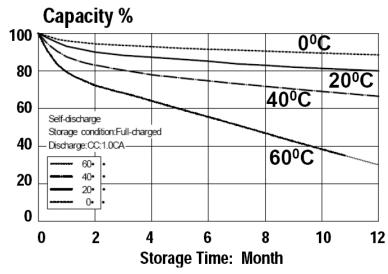


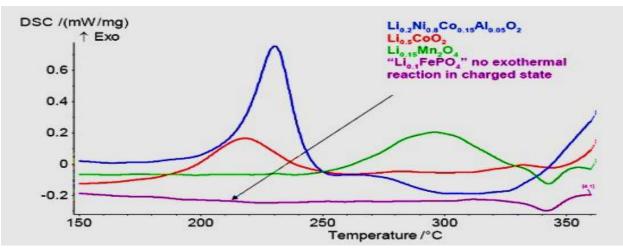
Factor to Cause Thermal Runaway in an Overcharge Event

- Material design is the most critical factor
 - LiCoO₂ and NMC type LIBs will usually result in thermal runaway after cell voltage is higher than 5.5V (without the aid of protection device(s))
 - LiMn₂O₄ type LIBs can possibly fail safely after cell voltage is higher than 10V
 - LFP type LIBs can possibly fail safely after cell voltage is higher than 15V

Temperature is Critical to LIB Performance and Safety



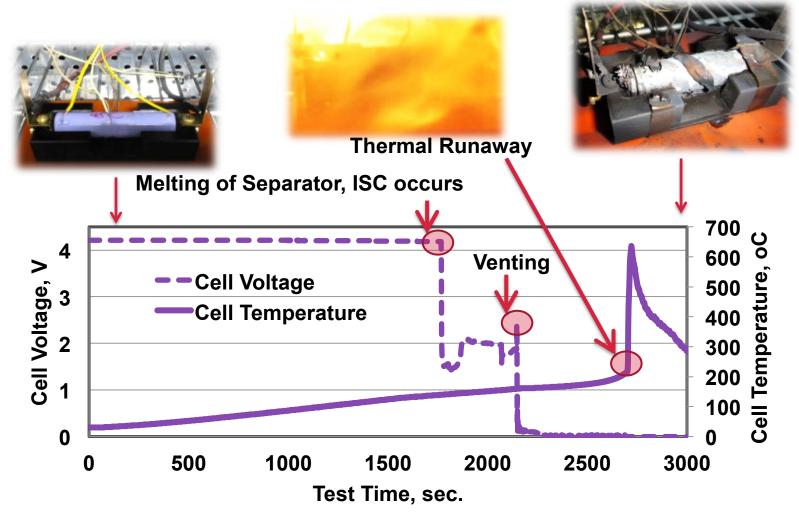




Source: G. Arnold et al. / Journal of Power Sources, 2003, 119–121, 247–251



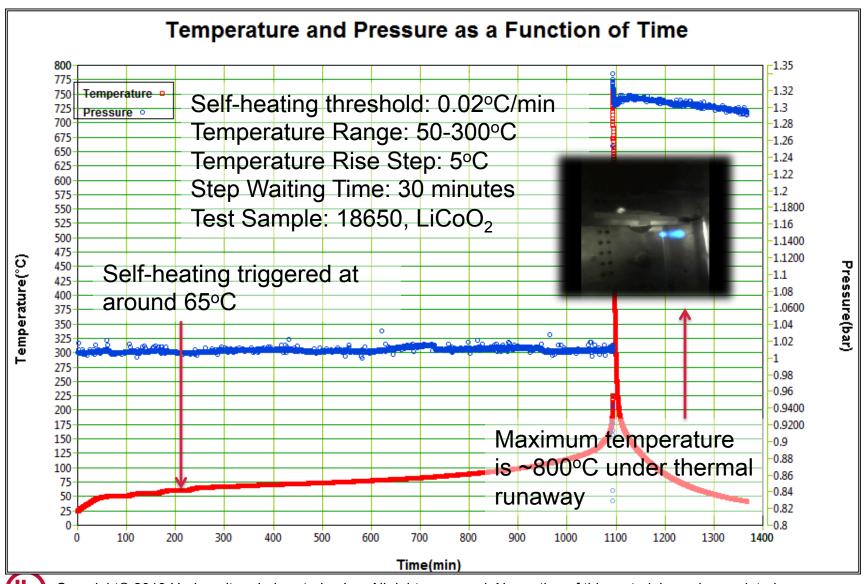
LIB Behavior(s) in Oven Heating Test



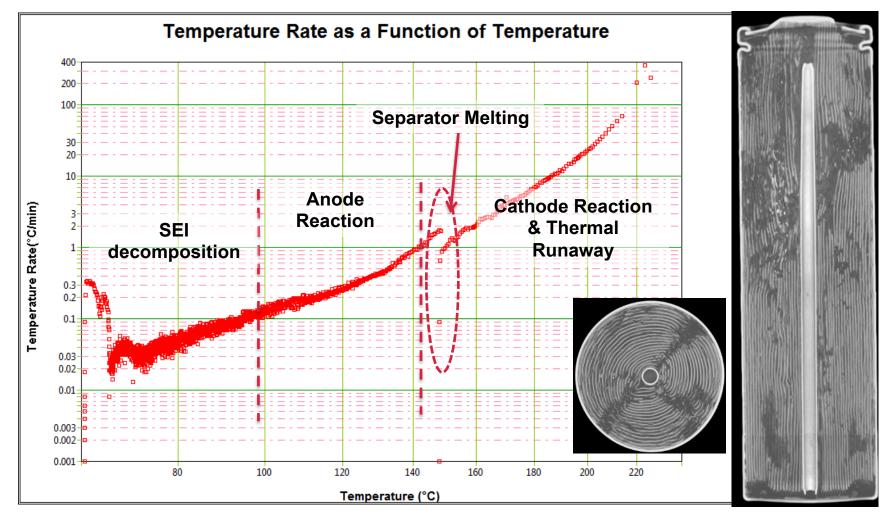
Heating Rate: 5°C/min; from room temperature to 180°C then keep 180°C until thermal runaway



LIB Behavior(s) in ARC Test – 18650 LiCoO₂

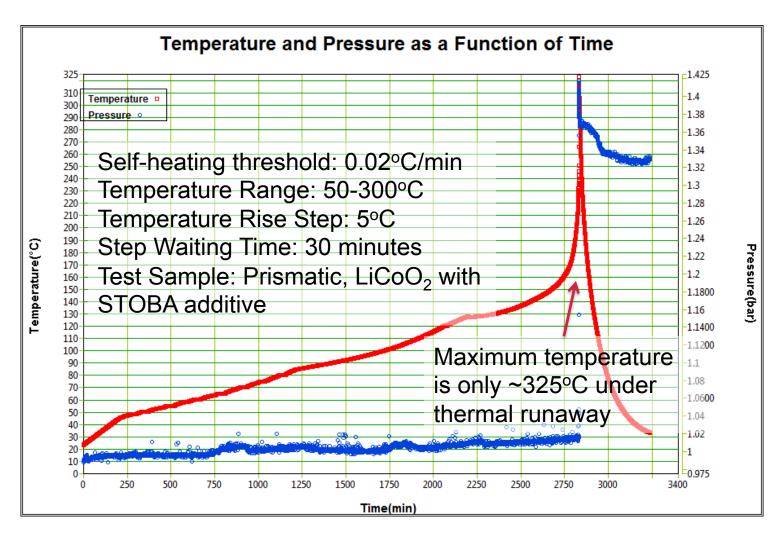


LIB Behavior(s) in ARC Test – 18650 LiCoO2 (cont.)



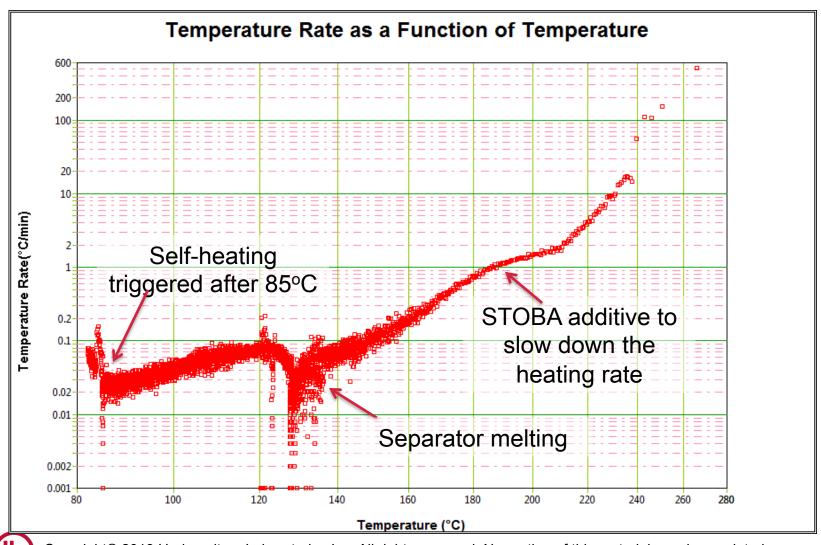


LIB Behavior(s) in ARC Test – Prismatic LiCoO2 with STOBA additive





LIB Behavior(s) in ARC Test – Prismatic LiCoO2 with STOBA additive (cont.)

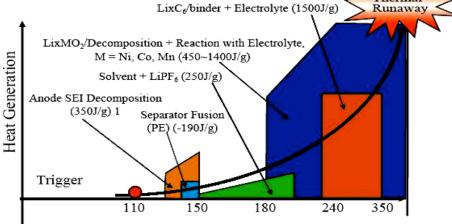


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Factor(s) to Cause Self-Heating Chain Reactions in a Thermal Abuse Event

- ◆ Poor SEI stability can trigger self-heating at lower temperature (<70°C)</p>
 - The compatibility between anode and electrolyte is critical
- SEI decomposition mechanism is critical at early stage of overheating

 Cathode material is important to the severity of thermal runaway at later stage





Summary

Internal Short-Circuit

- A initial local heating can sometimes introduce a more severe condition because the electrode materials can be still at high chemistry activity at initial stage
- Heat generation and dissipation rate are the two key factors to make cell fail safely or thermal runaway

Overcharge

- Under medium to high voltage region (4.5-5.2V), electrolyte oxidation can occur at cathode and to result in cell swelling
- Under higher voltage (>4.8V), cathode material can possibly decompose to generate oxygen and result in thermal runaway

Thermal Abuse

- Thermal stability of SEI is important to minimize the risk of self-heating at early stage of overheating (60°C-100°C)
- Anode/electrolyte reactions can generate heat at middle stage of overheating (100°C-180°C)
- Separator melting can slow down the self-heating rate (120-160°C)
- Cathode decomposition can release oxygen and abundant heat at latter stage of overheating (>180°C)



Example(s) of Strategies to Enhance LIB Safety (Cell Level)

Thicker separator to minimize the risk of ISC and to slow down the heat generation under overheating with the temperature range 120°C-160°C

Embedded safety devices within cell to break the circuit before the cell voltage goes to a level with high safety risk

Additive (in electrolyte) to stabilize the SEI and electrolyte under overheating

Coating or fire retardant additives (in electrodes) to enhance the heat dissipation capability and/or to slow down the heat generation while ISC and overheating

Higher A/C ratio to minimize the risk of lithium plating under overcharge condition

Appropriate cathode material design to enhance the cell tolerance to overcharge



